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The nomenclature of 1-aminoalkylphosphonic acids and derivatives: evolution of the code system

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The approach for the unification of published proposals for the nomenclature and abbreviations of aminoalkylphosphonic acids and their derivatives is presented. Their modification was made on the basis of the IUPAC-IUB rules concerning the nomenclature and code system of proteinogenic amino acids. Our present proposal formulates the supplementary code and nomenclature system allowing unambiguous description of phosphonic analogs of proteinogenic amino acids, their analogs, homologs, metabolites, and derivatives including phosphonopeptides.

Key words: aminoalkylphosphonic acids, aminoalkylphosphonates, phosphonopeptides, nomenclature, code system

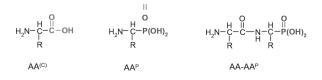
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INTRODUCTION

Aminophosphonic acids (AA^p) as structural analogs of protein amino acids (AA^c) are extremely important antimetabolites which compete with their carboxylic counterparts for the active sites of enzymes and other cell receptors (Kafarski & Mastalerz, 1984; Kukhar & Hudson, 2000; Kafarski & Lejczak, 2001).

Rapid development of chemistry of 1-aminoalkylphosphonic acids (AA^p) which took place in the 1970–1980 period (Kafarski & Mastalerz, 1984; Kafarski *et al.*, 1985; Kukhar & Hudson, 2000; Kudzin 2005; Lejczak & Kafarski, 2009; Orsini *et al.*, 2010; Kudzin *et al.*, 2011) — AA^p were the subject of more than 6000 papers published until 2001 (Kafarski & Lejczak, 2001) — generated a need for an update of the IUPAC–IUB nomenclature and code system (IUPAC–IUB, 1976–8), elaborated in the years 1976–1978 for phosphono–organics of biochemical importance (Table 1).

The first attempt for codification of aminophosphonic acids and their *N*-aminoacyl derivatives, namely P-terminal phosphonopeptides, was made as early as 1978 [L-Ala(P); L-Ala-L-Ala(P) (Allen *et al.*, 1978)]. This



attempt, based on the structural analogy of amino acids (2-aminoalkanoic acids; AA⁽⁵⁾) and amino-phosphonic acids (1-aminoalkylphosphonic acids; AA^{*p*}), adapted the IUPAC rules for the nomenclature and code system of proteinogenic amino acids. The code, proposed by Allen *et al.* (1978), allows for the differentiation between an amino acid and aminophosphonic acid, both alone and also as constituents of phosphonopeptides, e.g. Ala *vs.* Ala(P) (Table 2).

As a matter of fact, the substitution of the carboxylic function in an amino acid structure by the phosphonic function ($AA^{(c)} \rightarrow AA^{(c)}$) affords its phosphonic analog (Renaud & Seebach, 1986; Seebach *et al.*, 1989; Zecchini *et al.*, 1989; Corcoran & Green, 1990; Gerber & Seebach, 1991; Kaname *et al.*, 2001; Osapay *et al.*, 1987).



As a consequence of this fact, the substitution of long formal IUPAC names of 1-aminoalkylphosphonic acids, by short trivial names, derived from the amino acids nomenclature, was proposed. This approach expressed the structural analogy of aminophosphonic acids AA^p to their carboxylic analogs AA^(c), and was achieved by the addition of the "*phosphono*" prefix to the trivial names of amino acids (e.g., the phosphonic analog of alanine was named *phosphono*-alanine).

Thus, in 1980–1981 names with the prefix "phosphono" — phosphonohomocystine (Hcys^p)₂, phosphono-*S*-alkyl-homocysteine (RSHcys^p), and phoshonocystine (Cys^p)₂ — appeared in the literature (Kudzin & Stec, 1980; Kudzin, 1981). After these reports, the names of aminoalkylphosphonic acids with the *phosphono* prefix became more frequently used. The illustration of the evolution in aminophosphonate code and nomenclature, which took place during the last decades, is presented in Tables 2–4.

Modification of these proposals according to the IU-PAC-IUB rules (IUPAC-IUB, 1984) allowed for the

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Abbreviations: $\mathsf{AA}^\mathsf{p},$ 1-aminoalkylphosphonic acids; $\mathsf{AA}^\mathsf{c},$ 2-aminoalkanoic acids

No.	Structure	Name	Abbrev.
1	О II H ₃ C-C-O-Р(ОН) ₂	Acetyl phosphate	Ac-P
2	$ \begin{array}{c} $	β-Aspartyl phosphate	Asp(βP)
3		Carbamoyl phosphate	Cbm-P
4	$H_2 N - C - C - P(OH)_2$ $H_2 H_2 H_2$	(2-Aminoethyl)- phosphonic acid	
5	$\begin{array}{c} O & O \\ HO-P-C-U \\ OH \\ H_2 \end{array}$	(2-Oxoethyl)- phosphonic acid	P-CH₂CHO
6	$\begin{array}{c c} O & NH \operatorname{Me} & O \\ II & II & II \\ HO-P-N-C-N-C-C-OH \\ I & H \\ OH \end{array}$	Phosphocreatine	P-creatine
7	HO-P-N-C-NH2 OH	Phospho-guanidine	P-guanidine
8	$H_{2}N - C - C - O - P - O - C - O - P - O - C - O - P - O - C - O - O - O - O - O - O - O - O$	Aminoacyl adenylate [Jakubowski 1978, 1980, 1983, 1997]	AMP-Ser; Ser-P-Ado

Table 1. Names and abbreviations of some phosphorus-containing compounds of biochemical importance according to IUPAC-IUB rules (IUPAC-IUB, 1978)

formulation of the supplementary code and nomenclature system for phosphonic analogs of proteinogenic amino acids, and their analogs, homologs, metabolites, and derivatives, including phosphonopeptides. The guiding principles for this code and nomenclature system proposal are delineated in Tables 5–14.

PROPOSED NOMENCLATURE

Aliphatic Aminophosphonic Acids

Acyclic Aminophosphonic Acids

The fully systematic forms for acyclic aminophosphonic acids are derived from corresponding alkylphosphonic acids in which carbon linked to phosphorus atom is numbered as C1 or C α (Table 5). A heteroatom has the same number as the carbon atom to which it is attached, e.g. N-1 is on C-1, N-5 is on C-5, etc.

Phosphonic Analog of Proline

The carbon atoms in prolines, both in carboxylic as well as phosphonic, are numbered as in pyrrolidine, the nitrogen atom being numbered N-1, and proceeding towards the carboxyl or the phosphonic group (Table 6).

Aromatic Aminophosphonic Acids

The carbon atoms in aromatic rings of the phosphonic analogs of phenylalanine, tyrosine and/or tryptophan can be numbered as in systematic nomenclature, with 1' (or 3' for tryptophan), designating the carbon atom bearing the aliphatic chain. The carbon atoms of this chain can be designated with C1 or C α (for the carbon atom attached to the amino and phosphonic groups) and C2 or C β for the adjacent atom (the atom attached to the ring system) (Table 7).

Use of the Prefix 'Homo' and 'Nor'

An α -amino acid that is similar to one of the common ones, but that contains one more methylene group in the carbon chain, may be named by adding prefix 'homo' to the name of that common amino acid. '*Homo*' in the sense of a higher homologue is commonly used for homoserine and homocysteine.

The prefix 'nor' in the names 'norvaline' and 'norleucine' presents amino acidic isomers with not branched (normal) carbon chain. Both of these prefixes can be easily adapted for the nomenclature of aminophosphonates (Table 8 and 9).

Configuration at the α -Carbon Atom

This is usually described using the Fischer–Rosanoff convention (D- and L-) or Cahn-Ingold-Prelog (R- and S-) convention, in which L- α -amino acids, both the carboxyl (IUPAC CNOC, 1976) as well as the phosphonic forms (Dhavan & Redmore, 1987), are represented by the structures illustrated in Table 10.

These illustrate that L-amino acids and L-aminophosphonic acids, due to the order of preference: $\mathbf{CO}_{3}\mathbf{H} < \mathbf{N}\mathbf{H}_{2} < \mathbf{PO}_{3}\mathbf{H}_{2}$, exhibit different RS configurations. Thus, the L-configuration of amino acids (AA^(c)), nearly always corresponds to S (with the exception of L-cysteine), whereas the L-configuration of aminophosphonic acids (AA^{*i*}), always correspond to *R*.

The nomenclature "*allo*" and "*threo*" and also *cis* and *trans* have been applied in a similar way for the diastereoisomers of amino acids (threonines or prolines) and aminophosphonic acids (phosphonic analogs of threonine) (Bongini *et al.*, 1996).

Optical Rotation

The indication of the direction of rotation of plane polarized light (specified wavelength in a specified solvent), can be done with a 'plus' or 'minus' sign in parenthesis, e.g. (+)-alanine or (+)-phosphonoalanine [D-(+)-1-Aminoethylphosphonic acid; (S)-(+)-1-aminoethylphosphonic acid]. Subsequently, a racemic amino acid may be indicated by (±), e.g. (±)-leucine or (±)-phosphonoleucine.

SYMBOLISM — GENERAL CONSIDERATIONS ON THE THREE-LETTER SYMBOLS

The trivial name for aminophosphonic acids can be derived from common names of their carboxylic analogs (e.g., aminomethylphosphonic acid *vs.* glycine), modified

Table 2. Applied names and abbreviations for phosphonic analogs of proteinogenic amino acids and their metabolites and/or homologs with phospho or related prefix (phosphono or phospha)

Structure	$R-S - \begin{bmatrix} C \\ H_2 \end{bmatrix} = \begin{bmatrix} O \\ H_2 $	$\substack{\textbf{R}-\textbf{S}-\textbf{C}-\textbf{C}-\textbf{P}(\textbf{OH})_2\\\textbf{H}_2 \textbf{I}_2\\\textbf{NH}_2 \\ \textbf{NH}_2 \\ $	$ \begin{bmatrix} -s - \begin{bmatrix} c \\ H_2 \end{bmatrix}_2^2 \begin{bmatrix} 0 \\ -P(OH)_2 \end{bmatrix}_2^2 \\ NH_2 \end{bmatrix} $
Name	Phosphohomocysteine derivatives [Kudzin & Stec, 1980]	Phosphocysteine derivatives [Kudzin, 1981]	<i>Phosphohomocystine</i> [Kudzin & Stec, 1980; Tam <i>et al.,</i> 1982]
Abbrev.	RHcys ^P : Met ^P ; Eth ^P [Kudzin, 2005]	RCys ^P : MCys ^P ; Etcys ^P [Kudzin, 2005]	[Hcys ^P] ₂ [Kudzin, 2005]
Structure	$\begin{bmatrix} 0\\ -S-C-C-P(OH)_2\\H_2&I\\NH_2\end{bmatrix}_2$	$\begin{array}{c} O & O \\ HO - \overset{O}{\underset{II}{S}} + \overset{O}{\underset{I}{C}} \overset{O}{\underset{I}} \overset{O}{ } \overset{O}{\underset{I}} \overset{O}{\underset{I}} \overset{O}{\underset{I}} \overset{O}{\underset{I}} \overset{O}{ } \overset{O}{$	$\begin{matrix} 0 & 0 \\ HO - \begin{matrix} HO C - C - C - P (OH)_2 \\ HO - \begin{matrix} HO - C - C - P (OH)_2 \\ HO - C - C - C - P (OH)_2 \\ HO - C - C - C - P (OH)_2 \\ HO - C - C - C - C - P (OH)_2 \\ HO - C - C - C - P (OH)_2 \\ HO - C - C - C - C - P (OH)_2 \\ HO - C - C - C - C - C - C - C - C - C - $
Name	Phosphocystine [Kudzin, 1981]	Phosphonohomo-cysteic acid [Kudzin et al., 2005b]	<i>Phosphonocysteic acid</i> [Kudzin <i>et al.,</i> 2005b]
Abbrev.	[Cys [₽]]₂ [Kudzin, 2005]	Hcys ^P A [Kudzin <i>et al.,</i> 2005b]	Cys ^P A [Kudzin <i>et al.,</i> 2005b]
Structure	$HS = \begin{bmatrix} C \\ H_2 \end{bmatrix}_2^2 \begin{bmatrix} O \\ -P(OH)_2 \\ NH_2 \end{bmatrix}$	$\begin{array}{c} 0\\ H & I\\ HS - C - C - P(OH)_2\\ H_2 & I\\ NH_2 \end{array}$	$Me-S- \begin{array}{c} Me-S- \left(C_{H_{2}}^{-1} C_{2}^{H} C-P(OH)_{2} \right) \\ Me-S- \left(H_{2}^{-1} C_{2}^{H} C-P(OH)_{2} \right) \\ Me-S- \left(H_{2}^{H} C-P(OH)_{2} C-P(OH)_{2} \right) \\ Me-S- \left(H_{2}^{H} C-P(OH)_{2} C-P(OH)_{2}$
Name	Phosphohomocysteine [Kudzin & Stec, 1983]; Phosphonohomocysteine [Kudzin et al., 2005b; Beilstein base, 2005]	Phosphocysteine [Kudzin & Stec, 1983]; Phosphonocysteine [Kudzin <i>et al.</i> , 2005b; Beilstein base, 2005]	Phosphomethionine [Tam et al., 1982]
Abbrev.	[Hcys ^P]₂ [Kudzin <i>et al.,</i> 2005b]	[Cys ^P] ₂ [Kudzin <i>et al.,</i> 2005b]	Met [₽] [Kudzin, 2005]
Structure	$\begin{array}{c} X & H \\ Me - \overset{X}{\underset{H}{S}} + \overset{H}{\underset{H_{2}}{C}} \overset{H}{\underset{H_{2}}{S}} \overset{H}{\underset{H_{2}}{C}} + \overset{H}{\underset{H_{2}}{H}} (OH)_{2} \end{array}$	$\begin{array}{c} Me & O \\ I \leftarrow I \leftarrow I \leftarrow I \leftarrow I \\ Me - S \leftarrow I \leftarrow I \leftarrow I \leftarrow I \leftarrow I \leftarrow O \\ I \leftarrow I \leftarrow I \leftarrow I \leftarrow I \\ I \leftarrow I \leftarrow I \leftarrow I \\ NH_2 \end{array}$	$\substack{NH \\ R-\overset{II}{\underset{H}{\overset{II}{\underset{H_{2}}{\overset{II}{\underset{II}{\overset{II}{\underset{H_{2}}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\overset{II}{\underset{II}{\underset{II}{\overset{II}{\underset{II}{II}{\underset{II}{\atopII}}{\underset{II}{\underset{II}{\atopII}{\underset{II}{\atopII}{\underset{II}{\atopII}{\atopII}{\underset{II}{\atopII}{\underset{II}{\atopII}{\atopII}}{\underset{II}{\atopII}{\atopII}}{\underset{II}{\atopII}{\atopII}}{\underset{II}}{\underset{II}}{\underset{II}{\atopII}{\atop$
Name	Phosphomethionine sulfoxide (X=el. pair) or sulfon (X=O) [Tam et al., 1982]	Phosphomethionine methylsulfonium iodide [Tam et al., 1982]	Phosphohomocysteine sulfoximides [Nowakowski et al., 1989]
Abbrev.			
Structure	$H_2N - \begin{bmatrix} -C \\ H_2 \end{bmatrix} \stackrel{H}{\underset{n}{\overset{n}{\vdash}}} \stackrel{H}{\underset{n}{\overset{n}{\atop}}} \stackrel{H}{\underset{n}{\overset{n}{\atop}}} \stackrel{H}{\underset{n}{\overset{n}{\atop}}} \stackrel{H}{\underset{n}{\overset{n}{\atop}}} \stackrel{H}{\underset{n}{\overset{n}{\atop}}} \stackrel{H}{\underset{n}{\overset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\atop}} \stackrel{H}{\underset{n}{\underset{n}{}} \stackrel{H}{\underset{n}{\atop} \stackrel{H}{\underset{n}{}} \stackrel{H}{\underset{n}{} } \stackrel{H}{\underset{n}{\atop} \stackrel{H}{\underset{n}{}} \stackrel{H}{\underset{n}{} } \stackrel{H}{\underset{n}{} } \stackrel{H}{\underset{n}{$		$\begin{array}{c} HO-C-P(OH)_2\\ HO-C-P(OH)_2\\ HO-C-P(OH)_2\\ NH_2\end{array}$
Name	L-Phosphaornithine (n=3); L-Phosphalysine (n=4) [Wuggenig et al., 2011]	L-Phosphaproline [Wuggenig et al., 2011]	L-Phosphaserine [Hammerschmidt et al., 2000]
Abbrev.	Orn ^P ; Lys ^P [Kudzin, 2005]	Pro ^P [Kudzin, 2005]	L-Ser ^P [Kudzin, 2005]

Table 2. Continued	Table	2.	Continued
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Structure	$\begin{array}{c} O \\ H_2N - C - \overset{O}{C} - \overset{O}{P}(OH)_2 \\ H_2 \overset{O}{\overset{O}{P}} OH \end{array}$	Ме-С-Р(ОН) ₂ ОН NH ₂	$\begin{array}{c} O\\ Me-\overset{H}{\overset{H}{\underset{C}{\underset{C}{\overset{H}{\underset{C}{\underset{C}{\overset{H}{\underset{C}{\atop\atopC}{\underset{C}{\underset{C}{\underset{C}{\underset{C}{\atop\\{C}{\atop\\{C}{\underset{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{\atop\\{C}{{\atop\\{C}}{\atop\\{C}{{\atop\\{C}}{{{}}{{\atop\\{C}}{{\atop\\{C}}{{\\{C}}{{\atop\\{C}}{{\\{C}}{{\\{C}}{{{}}{{}}{{\\{C}}{{{}}{{}}$
Name	L- <i>Phosphaisoserine</i> [Hammerschmidt <i>et al.,</i> 2000]	Phosphothreonine [Bongini et al., 1994, 1996]	<i>lsothreonine-P</i> [Heisler <i>et al.,</i> 1995]
Abbrev.	L-iSer [₽] [Kudzin, 2005]	Thr [₽] [Kudzin, 2005]	isoTHR-P [Heisler <i>et al.,</i> 1995]
Structure	$\begin{array}{c} H_2 \ H_2$	$ \begin{array}{c} N \longrightarrow H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to}} H_2 \stackrel{O}{\underset{H}{\to} H_2 \stackrel{O}{\underset{H}$	$\begin{array}{c} \begin{array}{c} H_2 \\ H_2 \\ C \\ - \\ H \\ H \end{array} \\ H \\ H \end{array} \\ \begin{array}{c} H \\ H \\ H \\ H \\ H \\ H \\ H \end{array} \\ \begin{array}{c} H \\ H $
Name	<i>Phosphonohistidine</i> [Merrett <i>et al.</i> , 1988]	<i>Phosphonoisohistidine</i> [Merrett <i>et al.,</i> 1988]	<i>D,L-Phosphotryptophan</i> [Chen <i>et al.,</i> 1983; Beilstein base, 2005]
Abbrev.	His(P) [Merrett <i>et al.,</i> 1988]; His ^P [Kudzin, 2005]	isoHis(P) [Merrett <i>et al.,</i> 1988]; iHis ^P [Kudzin, 2005]	Trp ^P [Kudzin, 2005]
Structure	$\begin{array}{c} 1988]; His [Kudzin, 2005] \\ \hline 0 \\ HO \\ $	$\begin{array}{c} X - C - \begin{bmatrix} C \\ H_2 \end{bmatrix}_n \begin{bmatrix} C \\ H_2 \end{bmatrix}$	$\begin{array}{c} O & O \\ HO_{V} \overset{H}{=} \underbrace{C}_{P} \overset{H}{=} \underbrace{C}_{P} \overset{H}{=} \underbrace{U}_{C} \\ HO & NH_2 \end{array}$
Name	n=1: Phosphonoalanine [Tan & Tan, 1989; Beilstein base, 2005]; n=2: γ-Phosphonoglutamic acid [Mastalerz, 1957]		
Abbrev.	n=1: Asp ^{β-P} [Kudzin, 2005] n=2: Glu ^{γ-P} [Kudzin, 2005]	X=OH, n=1: Asp ^{α-P} [Kudzin, 2005] X=NH ₂ , n=1: Asn ^{α-P} [Kudzin, 2005] X=HO, n=2: Glu ^{α-P} [Kudzin, 2005] X=NH ₂ , n=2: Gln ^{α-P} [Kudzin, 2005]	n=1: Asp ^{P,P} [Kudzin, 2005; n=2: Glu ^{P,P} [Kudzin, 2005]
Structure	$\begin{array}{cccc} 0 & 0 \\ HO \\ P \\ P \\ HO \\ HO \\ HO \end{array} \begin{array}{c} 0 \\ HO \\ HO \\ HO \\ HO \end{array} \begin{array}{c} 0 \\ HO \\ $	$ \begin{array}{c} O\\ H_2N-C-C-P(OH)_2\\ H_2 H_2 \end{array} $	$ \begin{array}{c} 0 \\ H \\ C \\ -P(OH)_2 \\ H \\ NH_2 \\ OH \end{array} $
Name	N-Phosphono-glycine [Diel & Maier, 1984]; N-(Phosphonomethyl)- glycine [Aldrich, 2010; Beilstein base, 2010]		(o-hydroxyphenyl)- phosphonoglycine
Abbrev.	PMG	β-Ala ^ዖ [Kudzin, 2005]	OHPPG [Calvo, 1987]

by the addition of the prefix "*phosphono*" (aminomethyl-phosphonic acid \rightarrow phosphonoglycine).

Subsequently, the three-letter symbols of aminophosphonic acids (e.g., aminomethylphosphonic acid \rightarrow phosphonoglycine) can be formed by the modification of their carboxylic analog symbols (e.g., Gly for glycine) by the addition to their right side the capital letter symbolizing the phosphono group in the superscript (Gly *vs.* Gly^p). The comparison of nomenclature and symbolism for representative examples of amino acids and their phosphonic analogs is presented in Table 11.

Suggested nomenclature and symbolism for phosphonic analogs of the set of proteinogenic α -amino acids are listed in Table 12.

Table 3. Abbreviations for 1-aminophosphonic acids, phosphonopeptides, and derivatives

$\begin{array}{c} & R^{1} O \\ I & II \\ Y-N-C-P(OR^{*})_{2} \\ H & I \\ R \end{array}$	$\begin{array}{c} \begin{array}{c} O & O & O \\ Y-N-C-C-C-C-L-N-C-C-L-N-C-P(OR^{*})_{2} \end{array} \\ H & I_{1}^{1} \end{array}$	Literature
Y=H; R ¹ =H; R*=HO Ala(P); L-Ala(P); D-Ala(P)	Gly-L-Ala(P); Ala-L-Ala(P); Ala-D-Ala(P); Val-L- Ala(P); Phe-L-Ala(P); Pro-L-Ala(P); Met-L-Ala(P); Glu-L-Ala(P); Lys-L-Ala(P); (Ala) _n -Ala(P) (n=0-4)	[Allen <i>et al.</i> , 1978, 1979; Atherton <i>et al.,</i> 1979]
Y=H; R ¹ =H; R*=HO GlyP; LeuP; MetP; PheP; YR ¹ =-(CH ₂) ₃ -: ProP	Y=H; R*O=HO: Tyr-Gly-Gly-Phe-XP(OR) ₂	[Kupczyk- Subotkowska & Mastalerz, 1983]
	ValAlaP; AlaValP; ValGluP; LeuGluP; ValAdiP; ValLeuP	[Lejczak <i>et al.,</i> 1985]
	Z-Ala-"Ala"-OP(OPh) ₂ ; Z-Leu-"Ala"-OP(OPh) ₂ ; Z- Val-"Ala"-OP(OMe) ₂ ;	[Gerber & Seebach, 1991]
	[R*O=HO, EtO, iPrO]: L-Ala-L,D-AlaP; L-Leu-D,L- PheP; L-Leu-D,L-PheP]; L-Phe-L,D-LeuP; [L-Phe-L,D- LeuP] ¹ ; (Z)-L-Ala-L,D-AlaP(OiPr) ₂ ; (Z)-L-Val-L,D- AlaP(OiPr) ₂ ; (Z)-L-Phe-L,D-LeuP(OEt) ₂	[Galushko <i>et al.,</i> 1991, 1992]
Y=H; R ¹ =H; R*=HO Ala ^P ; D,L-Ala ^P Y=PhAc; R ¹ =H; R*=HO D,L-PhAc-Ala ^P		[Solodenko <i>et al.,</i> 1993]
Y=H; R^1 =H; R^*O =PhO Val. ^P (OPh) ₂ ; Phe. ^P (OPh) ₂	[Y=Boc; R*O=PhO]: Boc.Ala.Phe. ^P (OPh) ₂	[Hamilton <i>et al.,</i> 1993]
	[Y=H; n=0]: L-Ala-L,D-Ala P; L-Leu-D,L-Phe P; L- Phe-L,D-Leu P	[Heisler <i>et al.</i> , 1993]
Y=acyl (Ac, Prp, Btr, Piv, Bnz); R ¹ =H; R*=HO AA ^p : Gly ^p , Ala ^p , Hala ^p , Val ^p , Nval ^p , Nleu ^p , Met ^p , Pgly ^p , Phe ^p , Hphe ^p		[Kudzin <i>et al.,</i> 2005a]
(Y=TFA; R ¹ =H; R*=HO) TFA-AA ^P [AA ^P : Ala ^P , Val ^P , Pgly ^P , Phe ^P]		[Kudzin <i>et al.,</i> 2007]
(Y= mca; R ¹ =H; R*=HO) mca-AA ^P [AA ^P : Ala ^P , Val ^P , Pgly ^P , Phe ^P]	Gly-AA ^P [AA ^P : Ala ^P , Val ^P , Pgly ^P , Phe ^P]; MeGly-AA ^P [Ala ^P , Val ^P , Pgly ^P , Phe ^P]; Me ₂ Gly-AA ^P [Ala ^P , Val ^P , Pgly ^P , Phe ^P]; H ₂ N-Gly-AA ^P [Ala ^P , Val ^P , Pgly ^P , Phe ^P];	[Kudzin <i>et al.,</i> 2008]
AA ^P : full set of 24 amino acids	Atc-AA ^P	[Kudzin <i>et al.,</i> 2011]

Substituted Amino Acids

Substitutions in the Amino and Phosphonic Groups

The substitutions used in the amino acid chemistry (*N*-acylation, esterification, amidation), **transplanted** into the aminophosphonate chemistry are described by abbreviations listed in Table 13 and Table 14 (phosphonopeptides).

Thus, such substitutions can concern the amino function (i), the phosphonic function (ii) as well as both functions (iii) of aminophosphonic acid (AA_p) , which can be denoted as follows: — (i) the hyphen removes H from the 1-amino group of the AA^{p} when it is placed on the left of the symbol (- AA^{p});

— (ii) the hyphen removes OH from the 1-phosphonic group of the AAP (written in the conven-ional, not ionized form) when it is placed on the right of the symbol $(AA^{p}-)$;

— (iii) both modifications can apply to one symbol $(-AA^{P})$.

Table 4. Applied codes for representative phosphoramidate and phosphonamidate enzyme inhibitors

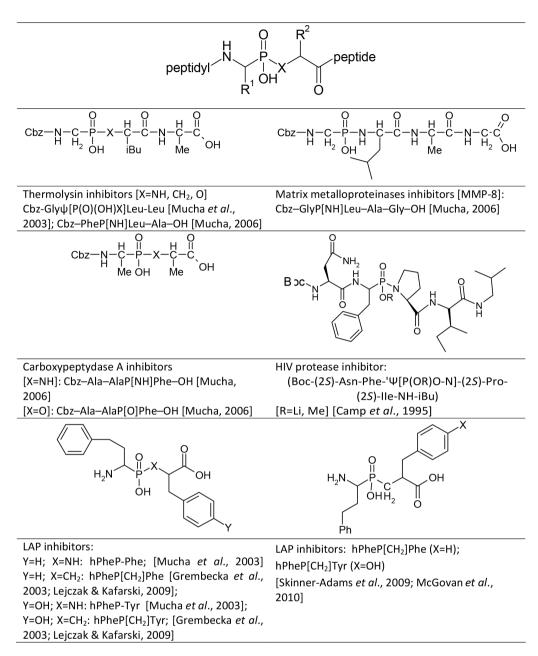


Table 5. Atom numbering in acyclic amino phosphonic acids

Parent	Structure of Parent Phosphonic	c Structure of Derived
Phosphonic	Acid	Aminoalkylphosphonic Acid
Acid		
$Me-PO_3H_2$	1 _Q	1 _Q
	H ₃ C-P(OH) ₂	$H_2 N - C - P(OH)_2$
	α	α
Names	Methylphosphonic acid	1-Aminomethylphosphonic acid
Et-PO ₃ H ₂	2 1 ₀	2 1 ₀ 2 1 ₀
	$H_3C-C-P(OH)_2$ H_2	$\begin{array}{ccc} H_{3}C - \begin{matrix} H & H \\ -C & -P(OH)_{2} \end{matrix} & H_{2}N - C - C - P(OH)_{2} \\ H_{2}N \end{matrix} \\ H_{2}N \end{array}$
	βα	βα

Names	Ethylphosphonic acid	1-Aminoethylphosphonic acid	2-Aminoethylphosphonic acid
Pr-PO ₃ H ₂	$\begin{array}{cccc} 3 & 2 & 1 \\ H_3C-C_2-C_2-P(OH)_2 \\ \gamma & \beta & \alpha \end{array}$	$\begin{array}{cccc} 3 & 2 & 1 \\ H_2N-C-C-C-C-P(OH)_2 \\ \gamma & \beta & \alpha \end{array}$	$\begin{array}{cccc} 3 & 2 & 1 \\ H_{3}C-C-C-C-P(OH)_{2} \\ H_{2} & / \alpha \\ \gamma & \beta & NH_{2} \end{array}$
Names	Propylphosphonic acid	1-Aminopropylphosphonic acid (GABA P-Analog)	3-Aminopropylphosphonic acid
Bu-PO ₃ H ₂ /Names	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	δ 1-Amir	$ \begin{array}{c} - & O \\ - & -C - C - P(OH)_2 \\ H_2 & H_2 & \alpha \\ NH_2 \end{array} $
	4-Aminobutylphosphonic aci	!-!	uanidinobutylphosphonic
Am-PO₃H₂	$5 4 3 2 1 0H_3C-C-C-C-C-C-P(OH)_2H_2 H_2 H_2 H_2 H_2$ $\epsilon \delta \gamma \beta \alpha$		$\begin{array}{cccc} 3 & 2 & 1 \\ C - C - C - P(OH)_2 \\ H_2 & H_2 \\ \gamma & \beta & \alpha \end{array}$
Names	Pentylphosphonic acid		pentylphosphonic acid

Table 6. Atom numbering in prolines

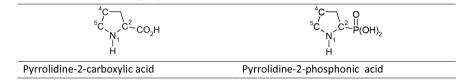


Table 7. Aromatic amino phosphonic acids

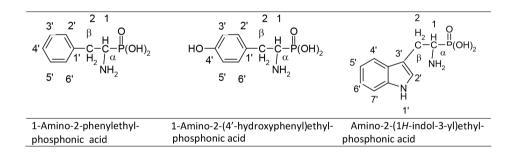
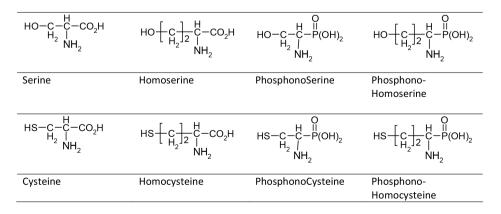


Table 8. Amino acids names with prefix "homo"





H_3C H HC-C-CO ₂ H H ₃ C H NH ₂	$H_3C \left[-C_{H_2} \right]_2 \left[-CO_2H_{H_2} \right]_2 \left[NH_2 \right]_3$	$\begin{array}{ccc} H_3C & O\\ H_3C - C - P(OH)_2\\ H_3C & I\\ NH_2 \end{array}$	$H_{3}C - \begin{bmatrix} C \\ -C \\ H_{2} \end{bmatrix} \begin{bmatrix} O \\ C \\ -P \\ OH)_{2} \\ NH_{2} \end{bmatrix}$
Valine	Norvaline	PhosphonoValine	PhosphonoNorvaline
$\begin{array}{c} H_{3}C, H \\ HC-C-C-C-C0_{2}H \\ H_{3}C, H_{2} H_{2} \\ NH_{2} \end{array}$	$H_3C - C_2H$ $H_3C - C_2H$ $H_2^3 - C_2H$ H_2	$\begin{array}{c} H_{3}C, & O\\ H_{2}C-C-C-P(OH)_{2}\\ H_{3}C, & H_{2}\\ \end{array}$	$H_{3}C - \begin{bmatrix} C \\ H_{2} \end{bmatrix}_{3} \begin{bmatrix} O \\ H \\ C \\ H_{2} \end{bmatrix}_{3} \begin{bmatrix} O \\ H \\ H_{2} \end{bmatrix}$
Leucine	Norleucine	PhosphonoLeucine	PhosphonoNorleucine

Table 10. Configuration comparison of proteinogenic amino acids and their phosphonic analogs

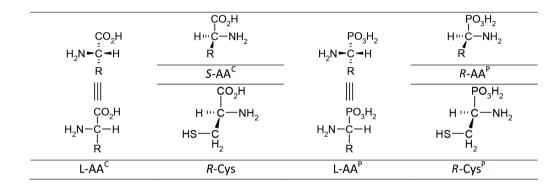


Table 11. Nomenclature and symbolism of amino acids and their phosphonic analogs

Amino Acid			Phosph	onic analog	
Structure	Name	Symbol	Structure	Name	Symbol
$H_2N-C-CO_2H$ H_2	Glycine	Gly	$\begin{array}{c} & O \\ H_2 N - C - P (OH)_2 \\ H_2 \end{array}$	Phosphono- Glycine	Gly ^P
H ₃ C-C-CO ₂ H NH ₂	Alanine	Ala	0 H ₃ C-С-Р(ОН) ₂ И NH ₂	Phosphono- Alanine	Ala ^P
$\begin{array}{c} H_2 N - C - C - C O_2 H \\ H_2 H_2 \end{array}$	2-Alanine; β-Alanine	2-Ala; β-Ala	$\begin{array}{c} & & \\ H_2N-C-C-P(OH)_2 \\ H_2 \ H_2 \ H_2 \end{array}$	Phosphono- 2-Alanine; Phosphono- β-Alanine	2-Ala ^P ; β-Ala ^P
$\begin{array}{c} HO-C-_{H_2}^{H} = CO_2H \\ H_2 = NH_2 \\ NH_2 \end{array}$	Serine	Ser	$\begin{array}{c} & & \\ & & \\ \text{HO-C-C-P(OH)_2} \\ & &$	Phosphono- Serine	Ser ^P

Table 12. Proposed nomenclature and symbolism for phosphonic analogs of proteinogenic α -amino acids

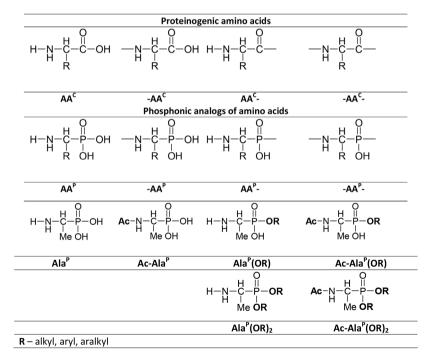
Trivial name ^a	Symb.	One- letter symb. ^b	Systematic name ^c	Formula ^c
Phosphono- Amino Acid	AA ^P	X ^P	1-Aminoalkylphosphonic acid	$H_2 N - C - P(OH)_2$
Phosphono- Glycine	Gly ^P	G ^P	Aminomethylphosphonic acid	$H_2 N - C - P(OH)_2 H_2$
Phosphono- Alanine	Ala ^P	A ^P	1-Aminoethylphosphonic acid	H II H ₂ N-C-P(OH) ₂ CH ₃
Phosphono- Valine	Val [₽]	V ^P	1-Amino-2-methylpropyl- phosphonic acid	Me O H II HC-C-P(OH) ₂ Me H ₂
Phosphono- Leucine	Leu ^P	L ^P	1-Amino-3-methylbutyl- phosphonic acid	$\begin{array}{c} & \overset{2}{\underset{\substack{\text{HC} \\ \text{HC} \\ \text{HC} \\ \text{Me} \\ \end{array}}{\overset{1}{\underset{\substack{\text{H} \\ \text{H} \\ \text{H} \\ \end{array}}}}} & \overset{2}{\underset{\substack{\text{H} \\ \text{H} \\ \text{H} \\ \text{H} \\ \text{H} \\ \end{array}}} \\ \begin{array}{c} & \overset{2}{\underset{\substack{\text{H} \\ \text{H} \\$
Phosphono- Isoleucine	lle [₽]	۱ ^Р	1-Amino-2-methylbutyl- phosphonic acid	$H_{3}C - C - C - C - C - C - C - C - O(OH)_{2}$ $H_{2} H - H - H_{2}$ $H_{3} H - H_{2}$
Phosphono- Phenylalanine	Phe ^P	F ^P	1-Amino-2-phenylethyl- phosphonic acid	$ \underbrace{ \begin{array}{c} & & \\ & & \\ & \\ & \\ & \\ & \\ & \\ & \\ & $
Phosphono- Proline	Pro ^P	P ^P	Pyrrolidin-2-yl-phosphonic acid	N P(OH) ₂
Phosphono- Serine	Ser ^P	S ^P	1-Amino-2-hydroxyethyl- phosphonic acid	$\begin{array}{c} & O \\ H \\ H O - C - \begin{array}{c} H \\ - C \\ - P \\ O \\ H_2 \end{array} \begin{array}{c} O \\ - P \\ O \\ N \\ N \\ N \\ \end{array} \right)$
Phosphono- Threonine	Thr ^P	T ^P	1-Amino-2-hydroxypropyl- phosphonic acid	$\begin{array}{c} OH & O\\ I & H & II\\ H_3C - \overset{C}{C} - \overset{C}{C} - P(OH)_2\\ H & I\\ NH_2 \end{array}$
Phosphono- Tyrosine	Tyr ^P	Y ^p	1-Amino-2-(4'-hydroxyphenyl)eth phosphonic acid	$HO \xrightarrow{- C - P(OH)_2} HO \xrightarrow{- U - P(OH)_2} HO \xrightarrow$
Phosphono- Cysteine	Ala ^P	A ^P	1-Amino-2-mercaptoethyl- phosphonic acid	$\substack{\textbf{HS-C-H}\\ \textbf{HS}-C-H_2 \\ \textbf{H}_2 \\ \textbf{H}_2 \\ \textbf{NH}_2 \\$
Phosphono- Methionine	Met ^P	M ^P	1-Amino-3-(methylthio)propyl- phosphonic acid (1-amino-3-methylsulfanyl- propylphosphonic acid)	$H_3C-S+C_{H_2}+C_2+P(OH)_2$
Phosphono- Lysine	Lys ^P	Κ ^Ρ	1,5-Diaminobutylphosphonic acid	$\begin{array}{c} H_{2}N + \begin{array}{c} H_{2} \\ H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} \begin{array}{c} H_{2} \\ H_{2} \end{array} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} \begin{array}{c} H_{2} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} \begin{array}{c} H_{2} \end{array} \end{array} $
Phosphono- Arginine	Arg ^P	R ^P	1-Amino-4-guanidinobutyl- phosphonic acid	$\begin{array}{c} \overset{NH}{\underset{H_2N-C-H}{\overset{H}{\underset{H_2}}} \overset{T}{\underset{H_2}} \overset{H}{\underset{H_2}} \overset{H}{\underset{NH_2}} \overset{O}{\underset{NH_2}} \\ \end{array}$
Phosphono- Histidine	His ^P	Η ^ρ	1-Amino-2-(1H-imidazol-4-yl)- ethylphosphonic acid)	$(1) \qquad \qquad$
Phosphono- Tryptophan	Trp ^P	W ^P	1-Amino- 2-(1H-indol-3-yl)ethyl- phosphonic acid	$ \underbrace{ \begin{array}{c} \begin{array}{c} & & \\ & & \\ & & \\ \end{array} \end{array} } \begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array} \end{array} \\ \begin{array}{c} & & \\ & \\ & \\ & \\ & \\ \end{array} \\ \begin{array}{c} & & \\ & \\ & \\ & \\ & \\ \end{array} \\ \begin{array}{c} & \\ & \\ & \\ \end{array} \\ \begin{array}{c} & \\ & \\ & \\ & \\ \end{array} \\ \begin{array}{c} & \\ & \\ & \\ & \\ & \\ & \\ \end{array} \\ \begin{array}{c} & \\ & \\ & \\ & \\ & \\ & \\ \end{array} \\ \begin{array}{c} & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & $

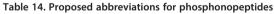
Table 12. Continued

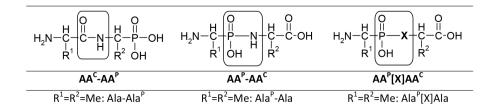
β-Phosphono- Aspartic Acid	Asp ^{βP}	D ^{βP}	2-Amino-3-phosphono- propionic acid	$\begin{array}{c} O & O \\ II & II \\ HO-C-C-C-C-P(OH)_2 \\ I & H_2 \\ NH_2 \end{array}$
α-Phosphono- Aspartic Acid	Asp ^{αP}	D ^{αP}	3-Amino-3-phosphono- propionic acid	$\begin{array}{c} 0 & 0 \\ 11 & 0 \\ HO-C-C-C-C-P(OH)_2 \\ H_2 & I \\ NH_2 \end{array}$
Phosphono- Aspargine	Asn ^{αP}	$N^{\alpha P}$	3-Amino-3-phosphono- propionamide (1-amino-2-carbamoyl-ethyl)- phosphonic acid	$\begin{array}{c} O & O \\ II \\ H_2N-C-C-C-C-P(OH)_2 \\ H_2 & I_2 \\ H_2 \\ NH_2 \end{array}$
α-Phosphono- Glutamic Acid	Glu ^{γP}	Ε ^{γΡ}	2-Amino-4-phosphono- butyric acid	$\begin{array}{c} O \\ HO - C - C - C \\ I \\ NH_2 \end{array} \begin{array}{c} O \\ O \\ I \\ P \\ O \\ P \\ O \\ P \\ O \\ O \\ P \\ O \\ O$
α-Phosphono- Glutamic Acid	Glu ^{αP}	Ε ^{αΡ}	4-Amino-4-phosphono- butyric acid	$\begin{array}{c} 0 \\ 1 \\ HO \\ -C \\ - \\ C \\ - \\ H_2 \\ - \\$
Phosphono- Glutamine	Gln ^{αΡ}	Q ^{ap}	4-Amino-4-phosphono- butyramide (1-amino-3-carbamoyl-propyl)- phosphonic acid	$\begin{array}{c} O \\ H_2 N - C - \left[- C \\ H_2 \right] \frac{1}{2} C \\ H_2 \\ I \\ I \\ N \\ N \\ H_2 \end{array} \begin{array}{c} O \\ I \\ I \\ N \\ H_2 \end{array}$

^aThe trivial name refers to the L or D or DL-amino acid. ^bUse of the one-letter symbols should be restricted to the comparison of long sequences (3-20 residues). ^cThe systematic names and formulas given refer to hypothetical forms in which amino groups are not protonated and phosphonic groups are not dissociated.

Table 13. Proposed abbreviations for phosphonic amino acids and derivatives







CONCLUSIONS

We believe that the implementation of the nomenclature and abbreviations proposed here will remove the ambiguity and will simplify the description of phosponate analogs of amino acids and their derivatives, including phosphonopeptides. Acceptance of these rules will facilitate unambiguous communication of research involving these classes of compounds.

Acknowledgements

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